

The art of purity: Why the future of rare earth metal metallurgy depends on the quality of quartz crucibles

Aleksandr Drozdov *

Expert in the field of glassblowing

World Journal of Advanced Research and Reviews, 2025, 25(03), 2530–2533

Publication history: Received on 18 February 2025; revised on 25 March 2025; accepted on 27 March 2025

Article DOI: <https://doi.org/10.30574/wjarr.2025.25.3.0965>

Abstract

The article is devoted to the study of the interrelation between the quality of crucibles made of fused quartz and the purity of the final product in the metallurgy of rare earth metals (REMs). The aim of the work is the scientific substantiation of the critical necessity of using high-purity quartz products to prevent contamination of REM melts and to ensure the specified functional properties of the materials. The article analyzes the physicochemical properties of fused quartz, considers the main types of impurities and structural defects that arise during the production of crucibles, and their influence on the performance characteristics of REM-containing alloys and compounds. It is substantiated that manufacturing craftsmanship in the fabrication of crucibles is a determining factor of technological success in this industry. The results can be applied by process engineers and materials scientists for developing specifications for consumables and for optimizing processes for obtaining high-purity REMs.

Keywords: Quartz Crucible; Fused Quartz; Rare Earth Metals; High-Purity Metallurgy; Alloying; Micro-Impurities

1 Introduction

Rare earth metals represent a group of 17 chemical elements possessing unique magnetic, luminescent, and catalytic properties. They are indispensable components in the production of high-technology products, including permanent magnets, lasers, catalysts, and batteries. The relevance of the study is due to the fact that the functional characteristics of final products directly depend on the purity of the REMs used. Even slight impurities at the level of parts per million (ppm) can drastically change the properties of materials. The processes of melting and casting REMs proceed at high temperatures and under conditions of high chemical activity of the melt, which imposes extreme requirements on container materials. The purpose of this article is to analyze and systematize the requirements for quartz crucibles and to prove that the quality of their manufacture is one of the fundamental conditions for the development of REM metallurgy.

* Corresponding author: Aleksandr Drozdov

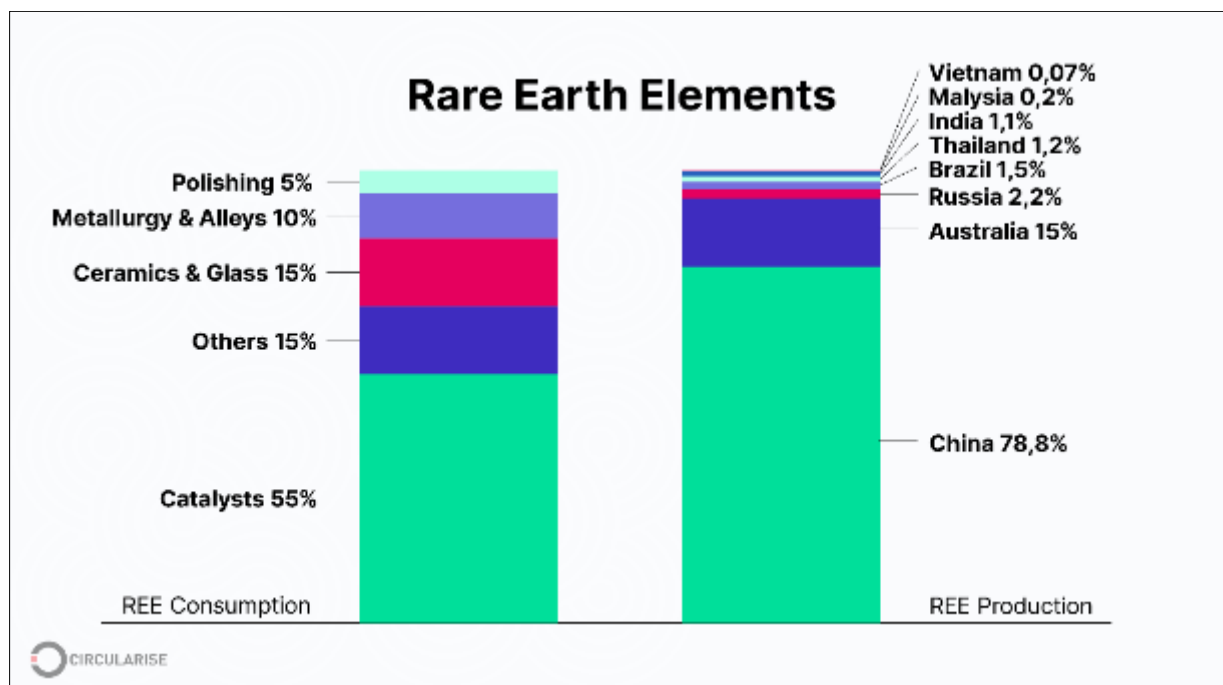


Figure 1 Rare Earth elements

1.1 Physicochemical Prerequisites for the Use of Fused Quartz

The choice of material for a crucible in REM metallurgy is dictated by a combination of unique conditions: the high melting temperatures of the metals (from 824 °C for cerium to 1663 °C for lutetium) and their exceptional chemical activity. Rare earth metals, especially in the molten state, actively interact with most elements, particularly with oxygen, carbon, and nitrogen [1]. This circumstance excludes the use of many traditional refractory materials, such as aluminum oxide or graphite, which either react with the melt or themselves become a source of its contamination. Fused quartz, which is an amorphous form of silicon dioxide (SiO_2), represents an almost optimal material for these tasks. Its softening temperature exceeds 1600 °C, which allows working with most REMs. Its exceptionally low coefficient of thermal expansion ensures high resistance to thermal shock, preventing crucible cracking during rapid heating and cooling. However, the most valuable property of quartz is its chemical inertness. Silicon dioxide possesses high thermodynamic stability, which minimizes its interaction with aggressive REM melts and prevents the transfer of silicon and oxygen into the final product [4].

1.2 Influence of Impurities in the Crucible Material on REM Properties

Despite its generally high purity, industrial fused quartz always contains some amount of trace impurities. Their source is the starting raw material natural vein quartz or synthetic silicon dioxide. The most common impurities are oxides of aluminum (Al_2O_3), iron (Fe_2O_3), alkali metals (Na_2O , K_2O), and hydroxyl groups (OH^-). Under high-temperature conditions, these impurities may diffuse from the crucible wall into the REM melt, causing its contamination. The consequences of such contamination can be critical. For example, iron impurities in neodymium magnets (NdFeB) reduce their coercivity and energy product. Aluminum, when entering phosphors based on europium or terbium, can change their spectral characteristics and decrease quantum yield. According to studies in REM metallurgy, controlling the content of boron and phosphorus at the sub-ppm level is mandatory for obtaining materials with specified magnetic properties [2]. Thus, even an insignificant number of impurities transferred from the crucible can completely devalue the expensive final product. This forms the strictest requirements for the purity of the initial quartz raw material used for the production of crucibles.

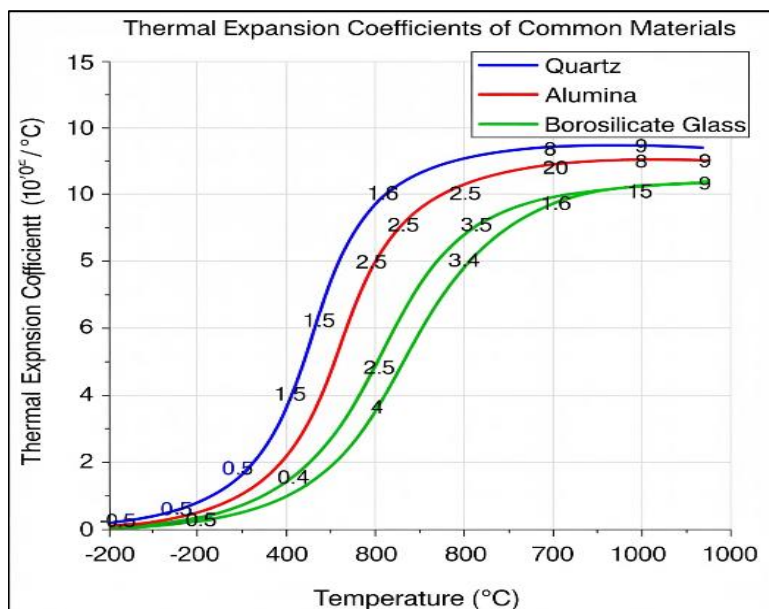


Figure 2 Thermal Expansion Coefficients of Common Materials

1.3 The Role of Structural Defects and Manufacturing Execution

In addition to chemical purity, physical homogeneity and the absence of structural defects in the crucible itself are of great significance. In the process of its fabrication, defects may arise such as gas bubbles, crystalline inclusions (cristobalite), and internal mechanical stresses.

Gas bubbles represent stress concentrators. At high temperatures and in contact with the melt, they can become points of crack initiation, which leads to mechanical failure of the crucible and loss of valuable material. Crystalline inclusions of cristobalite, which arise under improper temperature conditions during quartz processing, have a coefficient of thermal expansion that differs from that of the amorphous matrix. This creates local stresses that can also cause the crucible to crack during thermal cycling [3].

The quality of manufacture, therefore, directly influences the reliability and durability of the crucible. The professionalism of the glassblower is manifested in the ability to select optimal heating and forming regimes, to ensure uniform wall thickness, and to carry out proper annealing of the product to relieve internal stresses. Only such an approach guarantees that the crucible will withstand extreme operating conditions without destruction and will not become a source of secondary contamination of the melt.

The analysis conducted demonstrates that the quality of quartz crucibles is not a secondary but a fundamental factor determining success and economic efficiency in the metallurgy of rare earth metals. The future of this strategically important industry, which supports the development of electronics, “green” energy, and the defense sector, depends on the ability to produce REMs of the highest purity.

Achieving this purity is impossible without the use of container materials that do not introduce even a minimal amount of contaminants into the melt. High-purity fused quartz, free of structural defects, fully meets these requirements. The practical implementation of these conditions places a special responsibility on manufacturers of quartz products. Their expertise in selecting raw materials, craftsmanship in processing the material, and strict quality control at all stages of production become an integral part of the technological chain of producing modern high-technology REM-based materials.

2 Conclusion

The analysis conducted in this study demonstrates that the purity and structural integrity of fused quartz crucibles are critical determinants of both the technological success and economic efficiency of rare earth metal metallurgy. High-temperature interactions between REM melts and crucible materials make even trace-level impurities and structural defects unacceptable, as they can significantly alter the properties of the final products. Ensuring the use of high-purity fused quartz and flawless manufacturing practices minimizes contamination risks, increases process reliability, and

enhances the quality of REM-based materials. This research highlights the necessity of rigorous material selection, skilled craftsmanship, and comprehensive quality control throughout crucible production. By improving the standards of quartz crucible fabrication, this study contributes to advancing high-purity metallurgy, supporting the development of cutting-edge technologies, and ultimately benefiting society through more efficient, sustainable, and reliable production of strategic rare earth materials.

References

- [1] Gupta C. K., Krishnamurthy N. Extractive Metallurgy of Rare Earths. – 2nd ed. – Boca Raton: CRC Press, 2015. – 786 p.
- [2] Isshiki M. Purification of Rare Earth Metals // High Purity Metals / ed. by Y. Waseda, M. Isshiki, H. S. Chen. – Berlin, Heidelberg: Springer, 2001. – 365 p.
- [3] M. Khalifa, M. Hajji, H. Ezzaouia. Impurity removal process for high-purity silica production by acid leaching. – EDP Sciences, 2012.
- [4] Deal B.E., Helms C.R. The Physics and Chemistry of SiO₂ and the Si-SiO₂ Interface 2. – Springer, 1993. – 519 p.